

40. Delphinium Alkaloids. Part II. Ajacine.

By JOHN A. GOODSON.

Ajacine yields as products of hydrolysis lycoctonine and acetic and anthranilic acids and hence has the formula $C_{34}H_{46}O_9N_2 \cdot 2H_2O$.

AJACINE was first isolated from the seeds of *Delphinium ajacis* by Keller and Völker (*Arch. Pharm.*, 1913, 251, 209), who assigned to it the formula $C_{15}H_{21}O_4N \cdot H_2O$. Recently, Hunter (*Pharm. J.*, 1943, 150, 82) proposed the formula $C_{32}H_{44}O_8N_2 \cdot 2H_2O$.

As it was thought that ajacine might be closely related to lycaconitine from *Aconitum lycoctonum*, which Schulze and Bierling (*Arch. Pharm.*, 1913, 251, 8) showed to be succinylanthranoyl-lycoctonine, and to methyl-lycaconitine from *D. elatum*, which Goodson (*J.*, 1943, 139) found to be methylsuccinylanthranoyl-lycoctonine, it has been submitted to hydrolysis. Alkaline hydrolysis gives acetylanthranilic acid and lycoctonine according to the equation $C_{34}H_{46}O_9N_2 + H_2O = C_{25}H_{39}O_7N + C_9H_9O_3N$ and acid hydrolysis yields acetic acid and anthranoyl-lycoctonine: $C_{34}H_{46}O_9N_2 + H_2O = CH_3 \cdot CO_2H + C_{32}H_{44}O_8N_2$. From this it follows that ajacine is acetylanthranoyl-lycoctonine and has the empirical formula $C_{34}H_{46}O_9N_2 \cdot 2H_2O$.

EXPERIMENTAL.

The m. p.'s are corrected.

Ajacine crystallised from the concentrated ethereal solution of the alkaloids liberated by sodium bicarbonate from the solution of the total alkaloids of *D. ajacis* seeds in dilute hydrochloric acid. The base, purified by crystallisation from 70% alcohol, gave needles with two molecules of water, m. p. 154° (sintering at 150°), $[\alpha]_D^{25} + 49 \cdot 5^\circ$ ($c = 2$ in dry alcohol), $[\alpha]_D^{25} + 30 \cdot 8^\circ$ ($c = 2$ in *N*/5-hydrochloric acid) (Found: C, 61.5; H, 7.8; N, 4.5; OMe, 17.5; NMe, 4.0; loss at 105° in a vacuum, 5.4. $C_{34}H_{46}O_9N_2 \cdot 2H_2O$ requires C, 61.6; H, 7.6; N, 4.2; 4OMe, 18.7; NMe, 4.4; $2H_2O$, 5.4%).

Alkaline Hydrolysis of Ajacine.—Ajacine (5 g.) was dissolved in alcohol (125 c.c.) and heated on the water-bath with 17 c.c. of *N*-sodium hydroxide until most of the alcohol had evaporated. On addition of water (125 c.c.), 2.54 g. of crude crystalline lycoctonine separated. A further 0.71 g. was extracted from the filtrate by chloroform. Crude crystalline acetylanthranilic acid (0.71 g.) separated on the addition of dilute sulphuric acid; a further 0.69 g. was obtained by extraction of the filtrate with chloroform. The lycoctonine, purified by recrystallisation from 70% alcohol, had $[\alpha]_D^{25} + 53 \cdot 0^\circ$, m. p. 141°, not depressed by lycoctonine obtained from the hydrolysis of methyl-lycaconitine from the seeds of *D. elatum* (Goodson, *J.*, 1943, 141) (Found: equiv., by titration with *N*/10-sulphuric acid with methyl-red as indicator, 485. Calc. for $C_{25}H_{39}O_7N \cdot H_2O$: equiv., 483).

The acetylanthranilic acid, recrystallised from boiling water, had m. p. 185°, not depressed by authentic acetylan-

thranilic acid (Found : C, 60.6; H, 5.2; N, 7.9; equiv., by titration with $N/5$ -sodium hydroxide with phenolphthalein as indicator, 179.5. Calc. for $C_8H_9O_3N$: C, 60.3; H, 5.1; N, 7.8%; equiv., 179.1). Hydrolysis with dilute hydrochloric acid gave acetic and anthranilic acids.

Acid Hydrolysis of Ajacine.—Ajacine (5 g.) was dissolved in 50 c.c. of 10% hydrochloric acid and kept in a closed vessel for 14 days. On addition of excess of saturated solution of sodium carbonate, 4.7 g. of crude anthranoyl-lycoctonine were precipitated. The filtrate from the anthranoyl-lycoctonine was acidified with dilute hydrochloric acid and extracted with chloroform, which only removed a portion of the acetic acid produced. This was identified by conversion into the silver salt (Found : Ag, 64.3. Calc. : Ag, 64.6%). A further quantity of acetic acid was separated from the solution after extraction with chloroform as silver salt.

The anthranoyl-lycoctonine, purified by recrystallisation from 70% alcohol, had $[\alpha]_D^{25} + 31.0^\circ$ ($c = 2$ in $N/5$ -hydrochloric acid), m. p. 172° , not depressed by anthranoyl-lycoctonine obtained by hydrolysis of methyl-lycaconitine from *D. elatum*. Its solution had a beautiful blue-violet fluorescence. On alkaline hydrolysis it gave anthranilic acid and lycoctonine.

The author thanks Mr. F. J. Peters for assistance with the experimental work, and Messrs. A. Bennett and H. C. Clarke for the micro-analyses.

THE WELLCOME BUREAU OF SCIENTIFIC RESEARCH, LONDON, N.W.1.

[Received, December 29th, 1943.]